

Isothiazoles. Part X. New 3-Substituted-2-arylpropenamidines by Base Catalyzed Ring Opening of 3-Amino-4-arylisothiazole 1,1-dioxides.

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Abstract. Starting from 5-unsubstituted or 5-alkyl-, aryl-, heteroarylsubstituted 3-diethylamino-4-arylisothiazole 1,1-dioxides by base induced ring opening 3-alkoxypropenamidines were synthesized in excellent yields in a mild and efficient way. When 5-bromo-3-diethylamino-4-arylisothiazole 1,1-dioxide was used as the reagent, 3,3-dialkoxy-propenamidines, a new class of unsaturated amidines, were obtained. By using Grignard reagents 3-substituted acrylamidines can be produced.

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Keywords: Ring opening; Isothiazoles; Amidines; Sulfur heterocycles.

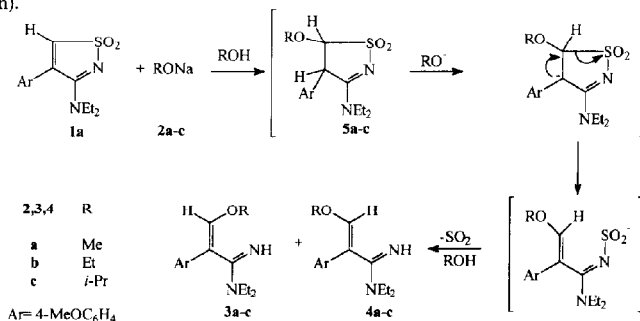
Introduction

Amidines are structural parts of numerous compounds of biological interest and form important medical and biochemical agents.^{1,2}

As known, amidines find widespread applications in organic synthesis especially for preparation of various heterocyclic systems. Although syntheses of amidines have been vigorously explored, few examples of unsaturated amidines (e.g. propenamidines) are reported in the literature and, particularly, of 3-hetero-substituted propenamidines.³⁻⁶ The interest in this field prompted us to explore the possibility of using the 3-amino-isothiazole 1,1-dioxide ring as an appropriate precursor of a new class of substituted propenamidines. In our laboratory we developed three main synthetic approaches to make available 3-amino-isothiazole 1,1-dioxides substituted at C-5.⁷⁻⁹ By these methodologies the preparation of derivatives unsubstituted or bearing at C-5 alkyl-, aryl-, heteroaryl-, vinyl- groups or substituents linked through an O- S- or N- atom is possible. In this paper we describe an efficient synthesis of 3- and 3,3-disubstituted 2-arylpropenamidines by base induced ring opening of 3-amino-4-arylisothiazole 1,1-dioxides. The availability of a great number of substituted 3-amino-4-arylisothiazole 1,1-dioxides, allowed the preparation both of a wide range of propenamidines with 3-alkyl-, aryl-, heteroaryl-, alkynyl-substitution which are not easy to synthesize by other methods and of several 3-alkoxy or 3,3-dialkoxy-propenamidines representing a new class of unsaturated amidines.

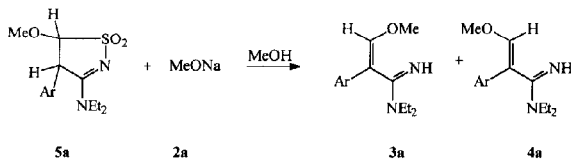
Results and Discussion

By reacting 3-diethylamino-4-(4-methoxyphenyl)-isothiazole 1,1-dioxide (**1a**) with a solution of two equivalents of sodium alkoxide (**2a-c**) in the corresponding alcohol at room temperature rapid disappearance of **1a** was observed. Upon work-up, the reaction mixture afforded in good yield (70-90%) two stereoisomeric products which were identified as 3-alkoxy-*N,N*-diethyl-2-(4-methoxyphenyl)-propenamidines **3a-c** (*Z* form) and **4a-c** (*E* form) in a ratio of about 3:1 respectively. Compounds **3b,c** and **4b,c** were obtained in mixture as uncrystallizable oils while pure **3a** could be isolated from the isomeric mixture by crystallizing with Et₂O. Pure samples of all isomers could be easily obtained by HPLC separation (see: Structure Characterization).



Scheme 1

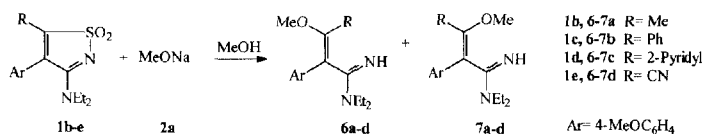
In a previous work we demonstrated that C-5 in the 3-amino-4-aryl-isothiazole 1,1-dioxide moiety is the more electrophilic carbon and nucleophiles attack at this position by a Michael-type reaction.¹⁰ It is known that 3-amino-4-arylisothiazole 1,1-dioxides easily react with alcohols in presence of a catalytic amount of the corresponding alkoxide and form *trans* and *cis* 5-alkoxy-4,5-dihydroisothiazole dioxide derivatives in a mixture where the *trans* isomer is the major one.⁹



Scheme 2

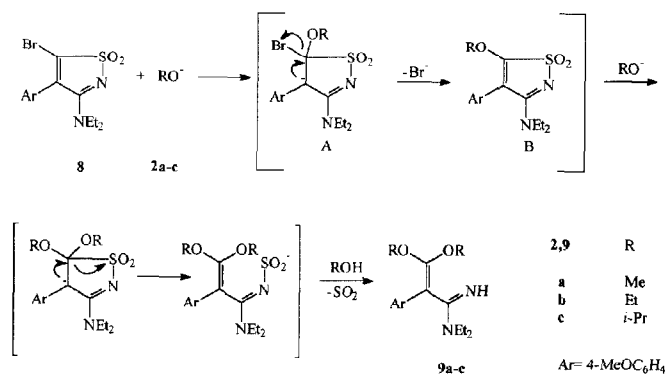
In the present case by using two equivalents of alkoxide we assumed that the 4,5-dihydro intermediate would react with another equivalent of the alkoxide with deprotonation at C-4. Ring opening of the anion and

spontaneous SO₂ elimination results in the formation of **3** and **4** (Scheme 1). This was confirmed by an independent experiment where 3-diethylamino-4,5-dihydro-5-methoxy-4-(4-methoxyphenyl)-isothiazole 1,1-dioxide **5a** was reacted with sodium methoxide in methanol at room temperature. Rapid disappearance of **5a** was observed and elaboration of the reaction mixture afforded **3a** and **4a** in a ratio of about 3:1. The results were the same, also in terms of isomer ratio, performing the reaction with sodium methoxide on the *cis* or on the *trans* or on the mixture of the two isomers (Scheme 2). These results prompted us to investigate the possibility to make this reaction more general to obtain 3-substituted-3-alkoxypropenamidines from 5-substituted isothiazole 1,1-dioxides. Starting from 3-diethylamino-4-(4-methoxyphenyl)-isothiazole 1,1-dioxides (**1b-e**) with the same procedure described above, 3-alkoxypropenamidines **6a-d** (*E* form) and **7a-d** (*Z* form), with different substituents on C-3, were synthesized. The *E* isomer was predominant possibly owing to the lower steric hindrance and in the case of compounds **a-c** only traces of the *Z* isomer **7** were detected (Scheme 3).



Scheme 3

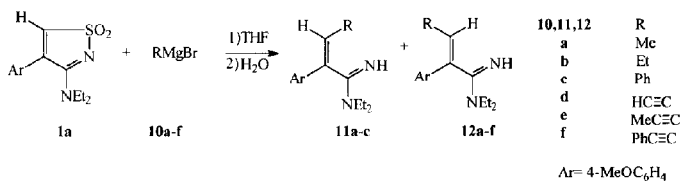
When the reaction using different alkoxides in alcohols was performed on 5-bromo-3-diethylamino-4-(4-methoxyphenyl)-isothiazole 1,1-dioxide (**8**), 3,3-dialkoxypropenamidines **9a-c** were formed. A reasonable mechanism for the formation of compounds **9** is depicted in Scheme 4.



Scheme 4

As an extension, we took into consideration the applicability of this ring opening reaction for the synthesis of the 3-alkyl, aryl, alkynyl-substituted propenamidines by reacting isothiazole 1,1-dioxide with Grignard reagents. Reaction of compound **1a** with **10a-f** in anhydrous THF at room temperature afforded in good yields the expected isomeric propenamidines **11a-c** and **12a-f**. It appeared reasonable that a reaction mechanism similar to that described above for the reaction between **1** or **8** and alkoxides was operative.

In the case of **11a-c** and **12a-c** a mixture of *E* and *Z* isomers in a ratio of about 2/3:1 was obtained. Only one isomer was observed performing the reaction between **1a** and **10 d-f** affording *E* isomer **12 d-f**. By this way highly conjugated systems of promising reactivity could be easily prepared (Scheme 5).



Scheme 5

Structure Characterization

¹H-, ¹³C-NMR and IR spectra were performed in order to characterize all compounds and to clarify, when needed, the configuration at the double bond. HPLC analyses of the isomeric mixtures were also performed using a CH₃CN : H₂O mixture in a 2:8 ratio as mobile phase with NaH₂PO₄ 1mg/mL and flow rate 0.7 mL/min. By this method analytical pure samples of each isomer or of the major one were obtained and analyzed by NMR techniques. To establish the configuration of the double bond we performed NOE difference and NOESY experiment observing the spatial relationship between the *o,o'* hydrogens of the 4-methoxy substituted aromatic ring on C-2 and the characteristic substituent on C-3, that is, for compounds **3/4a-c** and **6/7a-d**, the methoxy group and for **11/12a-f** the hydrogen or the alkyl group.

¹H-NMR of compounds **3a-c** and **4a-c** were mainly characterized by a sharp singlet in the range of 6.40-6.50 δ associated with H-3 and a broad singlet exchangeable with D₂O clearly associated with the NH group (IR NH 3300-3600 cm⁻¹). The signals associated with the *o,o'* hydrogens of the 4-methoxy substituted aromatic ring appeared to be shifted by 0.3-0.4 δ in one isomer of each pair with respect to the other. In the ¹³C-NMR spectra of **3a** and **4a** the chemical shift (55.70 δ) of the methoxy groups on the aromatic ring and of the methoxy groups on C-3 (60.8 δ) of each isomer appeared to be overlapped. A little difference (about 2 δ) in the chemical shifts are evident for C-3 and C-1. The double bond configuration of the *Z*-isomer **3a**, isolated from the reaction mixture through crystallization from Et₂O, was assigned by NOESY experiment showing a clear relationship between the aromatic protons (7.20 δ) and the hydrogen on C-3 (6.43 δ). The couple **3b/4b**, **3c/4c** were separated by HPLC and configuration was established through n.O.c experiments

as explained above. $^1\text{H-NMR}$ spectra of compounds **6/7** were mainly characterized by two singlets associated with the two methoxy groups and by the double doublets of the AX system associated with the hydrogens of the 4-methoxy substituted aromatic ring that always show a clear difference of chemical shift between the two isomers. It has to be noted that only one isomer was obtained pure in the case of **6/7 a-c**, only traces of the other isomer being detected by NMR and HPLC analyses. For **6a/7a** a small difference of the chemical shift of the methyl group on C-3 (*Z* isomer 1.97 δ ; *E* isomer 2.06 δ) was also observed. In the $^1\text{H-NMR}$ spectra of **6c/7c** characteristic signals of the aromatic pyridine hydrogens were observed. In the case of the couple **6,7d** the isomer **7d** could be isolated by crystallizing from Et_2O and the configuration at the double bond was assigned by NMR performing NOESY and NOE difference experiments: the methoxy signal at lower field could be associated with the methoxy group on the aromatic ring due to a clear n.O.e with the hydrogens of the 4-methoxy substituted aromatic ring on C-2 at 6.92 δ . The other methoxy group showed no effect with the aromatic ring thus confirming the *Z* configuration of **7d**. Compounds **9a-c** showed very simple $^1\text{H-}$ and $^{13}\text{C-NMR}$ spectra mainly characterized by the signals of two methoxy-, ethoxy or isopropoxy group respectively. The same procedure applied for **3/4** and **6/7** were also applied to establish the structure of compounds **11/12**. $^1\text{H-NMR}$ spectra of compounds **11a/12a** and **11b/12b** were mainly characterized by the signals associated with H-3 shifted by 0.2–0.3 δ in one isomer with respect to the other. In the $^{13}\text{C-NMR}$ spectra the signals associated with the methoxy groups of each isomer appeared overlapped. Instead, the singlets associated with C-3, C-1 and the aromatic quaternary carbons are separated by about 2 δ in each isomer. Only one isomer was obtained when the reaction with **1a** was performed with Grignard reagents **10d-f**. In all cases the configuration of the double bond of the more abundant *E* isomer was assigned by performing n.O.e experiments as explained above.

Conclusion

In conclusion we have demonstrated that 3-diethylamino-4-(4-methoxyphenyl)-isothiazole 1,1-dioxides are suitable materials for the synthesis of substituted 2-arylpropenamides through base induced ring opening. By this method two new classes of compounds, 3-alkoxy- and 3,3-dialkoxypropenamides could be synthesized. This reaction could be applied also for the synthesis of 2-arylpropenamides bearing alkyl-, alkynyl-, aryl substituents on C-3 and represents a good alternative to the known methods due to the mild reaction conditions and the high yields (70–90%).

Experimental

$^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ spectra were obtained with Bruker AC 200, Bruker AC 300 and Varian Gemini 200 instruments. Melting points were determined using a Buchi 510 (capillary) or a Electrothermal 9100 apparatus. Mass spectra were obtained by electron impact ionization at 70 eV from a Finningan INCOS 50

instrument using the direct exposure probe (DEP). IR spectra were recorded on a JASCO IR Report 100 spectrophotometer. HPLC analyses were performed on a Perkin Elmer LC 250 instrument, UV-visible detector LC 290, column Lichrocart 125-3 Purosphere RP18e (5 μ m). Mobile phase, CH₃CN : H₂O (2:8) NaH₂PO₄ 1mg/mL, flow rate 0.7 mL/min.

Materials

Compounds **1a-e**, **5** and **8** have already been described.⁷⁻¹¹ Grignard reagents **10a-c** are commercially available in 3.0 M sol. in Et₂O, **10d** in 1.0 M sol. in THF, **10e,f** in 0.5 M sol. in THF (Aldrich).

Synthesis of Propenamidines **3a-c** and **4a-c**:

Compound **1a** (0.15 g, 0.5 mmol) was dissolved in the appropriate alcohol (10 mL) and sodium alkoxide **2a-c** (5 mL of a 0.2 molar solution) was added under stirring at room temperature. When the reagent disappeared (T.I.C. cyclohexane/ethyl acetate 1:1; CH₂Cl₂/MeOH 10:1) water (5 mL) was added and stirring continued for 1 h. Alcohol evaporation and extraction with CH₂Cl₂ (10 mL) afforded the mixture of isomeric compounds **3a-c** and **4a-c**. Compounds **3b,c** and **4b,c** were obtained in mixture as oils while pure **3a** could be isolated from the isomeric mixture by crystallizing with Et₂O. Pure samples of all isomers could be easily obtained by HPLC separation.

N,N-Diethylamino-3-methoxy-2-(4-methoxyphenyl)-propenamidines **Z-3a**, **E-4a**. Total yield 85%. **3a** (white crystals): M.p. 86°C (Et₂O). IR (Nujol) 3100-3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 0.90-1.20 (m, 6H, CH₃); 3.20-3.50 (m, 4H, CH₂); 3.74 (s, 3H, OCH₃); 3.76 (s, 3H, OCH₃); 6.43 (s, 1H, H-3); 6.83 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.00-7.30 (bs, 1H, NH, exch.); 7.20 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 13.50, 42.36, 55.70, 60.87, 114.68, 119.31, 126.4, 127.24, 144.53, 158.73, 163.75. Calcd. for C₁₅H₂₂N₂O₂ (262.35): C 68.67, H 8.45, N 10.68 found C 68.64, H 8.37, N 11.10. *m/z* 262 (M⁺) **4a** (mixture with **3a**): ¹H-NMR (CDCl₃): 0.90-1.20 (m, 6H, CH₃); 3.10-3.40 (m, 4H, CH₂); 3.74 (s, 3H, OCH₃); 3.80 (s, 3H, OCH₃); 6.36 (s, 1H, H-3); 6.85 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.20-7.40 (bs, 1H, NH, exch.); 7.46 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 13.70, 42.20, 56.15, 61.20, 114.10, 118.52, 128.42, 129.50, 146.60, 159.00, 166.30. *m/z* 262 (M⁺).

N,N-Diethylamino-3-ethoxy-2-(4-methoxyphenyl)-propenamidines **Z-3b**, **E-4b**. Total yield 90%. IR (Nujol) 3100-3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 1.03-1.15 (m, 6H+6H); 1.29 (t, *J* 7 Hz, 3H); 1.34 (t, *J* 7 Hz, 3H); 3.10-3.50 (m, 4H+4H, NCH₂); 3.78 (s, 3H, OCH₃); 3.79 (s, 3H, OCH₃); 3.95 (q, *J* 7 Hz, 2H); 3.97 (q, *J* 7 Hz, 2H); 6.00-6.30 (bs, 1H, NH, exch.); 6.41 (s, 1H, H-3, *E* isomer); 6.43 (s, 1H, H-3, *Z* isomer); 6.82 (d, AB system, *J* 9 Hz, 2H, aryl-H); 6.85 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.20 (d, AB system, *J* 9 Hz, 2H, aryl-H, *Z* isomer); 7.48 (d, AB system, *J* 9 Hz, 2H, aryl-H, *E* isomer). ¹³C-NMR (CDCl₃): 13.39, 15.82, 42.91, 55.70, 69.54 (*E*), 70.06, 114.22 (*E*), 114.81, 126.43, 129.57 (*E*), 127.06, 128.16 (*E*), 144.16, 146.40 (*E*), 158.78, 159.18 (*E*), 163.61, 166.31 (*E*). Calcd. for C₁₆H₂₄N₂O₂ (276.38): C 69.53, H 8.75, N 10.14 found C 69.84, H 8.62, N 9.80. *m/z* 276 (M⁺). HPLC: *E* isomer r. t. 5.9 min., *Z* isomer r. t. 9.93 min.

N,N-Diethylamino-3-isopropoxy-2-(4-methoxyphenyl)-propenamidines **Z-3c**, **E-4c**. Total yield 70%. IR (Nujol) 3100-3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 1.04 (t, *J* 7 Hz, 6H+6H); 1.27 (d, *J* 6.2 Hz, 6H); 1.33 (d, *J*

6.2 Hz, 6H); 3.28 (q, NCH₂, J 7 Hz, 4H+4H); 3.79 (s, 3H, OCH₃); 3.80 (s, 3H, OCH₃); 4.07 (sept, J 6.2 Hz, 1H+1H); 5.80–6.01 (bs, 1H, NH, exch.); 6.45 (s, 1H, H-3); 6.47 (s, 1H, H-3); 6.83 (d, AB system, J 9 Hz, 2H, aryl-H, *E* isomer); 6.85 (d, AB system, J 9 Hz, 2H, aryl-H, *Z* isomer); 7.21 (d, AB system, J 9 Hz, 2H, aryl-H, *Z* isomer); 7.51 (d, AB system, J 9 Hz, 2H, aryl-H, *E* isomer). ¹³C-NMR (CDCl₃): 13.49, 22.94, 42.55, 55.63, 76.13 (*E*), 76.73, 114.10 (*E*), 114.70, 126.36, 129.48 (*E*), 117.41 (*E*), 118.26, 127.44 (*E*), 128.72, 142.56 (*E*), 144.60, 158.58 (*E*), 158.98, 164.04, 166.73. Calcd. for C₁₇H₂₆N₂O₂ (290.41): C 70.31, H 9.02, N 9.65 found C 70.64, H 8.87, N 9.40. (*E*). *m/z* 290 (M⁺). HPLC: *E* isomer r. t. 25.13 min., *Z* isomer r. t. 38.7 min.

Synthesis of Propenamides 6a-d and 7a-d:

Compounds **1b-e** (0.65 mmol) were dissolved in methanol (10 mL) and sodium methoxide **2a** (70.10 mg, 1.30 mmol) was added under stirring at room temperature. When the reagent disappeared (T.L.C. cyclohexane/ethyl acetate 1:1; CH₂Cl₂/MeOH 10:1) water (5 mL) was added and stirring continued for 1 h. Methanol evaporation and extraction with CH₂Cl₂ (20 mL) afforded compounds **6a-d** and **7a-d**. In the case of compounds **a-c** the *E* isomers **6** were obtained and only traces of **7** were detected by NMR and/or HPLC analyses. A mixture of **6d/7d** was obtained (**6/7** about 1:5 calculated by NMR) and pure **7d** isolated by crystallization with Et₂O.

N,N-Diethylamino-3-methoxy-2-(4-methoxyphenyl)-3-methylpropenamides *E*-**6a**, *Z*-**7a**. Total yield 90%. **6a** (pale yellow oil): IR (Nujol) 3100–3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃) 0.90–1.10 (m, 6H, CH₃); 1.97 (s, 3H, 3-CH₃); 3.20–3.40 (m, 4H, CH₂); 3.64 (s, 3H, OCH₃); 3.80 (s, 3H, OCH₃); 5.00–5.30 (bs, 1H, NH, exch.); 6.84 (d, AB system, J 9 Hz, 2H, aryl-H); 7.18 (d, AB system, J 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 16.54, 55.59, 113.80, 118.79, 128.05, 129.88, 150.19, 158.26, 166.87. Calcd. for C₁₆H₂₄N₂O₂ (276.38): C 69.53, H 8.75, N 10.14 found C 69.85, H 8.55, N 10.00. *m/z* 276 (M⁺). **7a**: (mixture with **6a**): ¹H-NMR (CDCl₃): 0.80 (t, J 7 Hz, 6H); 2.06 (s, 3H, 3-CH₃); 3.20–3.40 (m, 4H, CH₂); 3.63 (s, 3H, OCH₃); 3.80 (s, 3H, OCH₃); 5.55–5.90 (bs, 1H, NH, exch.); 6.82 (d, AB system, J 10 Hz, 2H, aryl-H); 7.44 (d, AB system, J 10 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 15.07, 55.70, 114.10, 130.48, 151.09, 158.73, 165.77.

N,N-Diethylamino-3-methoxy-2-(4-methoxyphenyl)-3-phenylpropenamides *E*-**6b**, *Z*-**7b**. Total yield 90%. **6b** (pale yellow oil): IR (Nujol) 3100–3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃) 0.80–1.14 (m, 6H, CH₃); 3.10–3.50 (m, 4H, CH₂); 3.43 (s, 3H, OCH₃); 3.73 (s, 3H, OCH₃); 6.10–6.50 (bs, 1H, NH, exch.); 6.65 (d, AB system, J 9 Hz, 2H, aryl-H); 6.99 (d, AB system, J 9 Hz, 2H, aryl-H); 7.15–7.35 (m, 5H, aryl-H). ¹³C-NMR (CDCl₃): 13.66, 42.20, 55.15, 58.24, 113.78, 122.08, 127.44, 128.45, 128.86, 130.16, 131.00, 133.00, 152.90, 158.50, 165.10. Calcd. for C₂₁H₂₆N₂O₂ (338.45): C 74.53, H 7.74, N 8.28 found C 74.94, H 7.37, N 8.11. *m/z* 338 (M⁺). **7b** (mixture with **6b**): ¹H-NMR (CDCl₃): 0.70–1.20 (m, 6H, CH₃); 3.00–3.50 (m, 4H, CH₂); 3.49 (s, 3H, OCH₃); 3.82 (s, 3H, OCH₃); 6.85 (d, AB system, J 10 Hz, 2H, aryl-H); 7.30–7.50 (m, 5H, aryl-H); 7.50 (d, AB system, J 10 Hz, 2H, aryl-H). HPLC: *E* isomer r. t. 4.99 min., *Z* isomer r. t. 7.4 min.

N,N-Diethylamino-3-methoxy-2-(4-methoxyphenyl)-3-(2-pyridyl)-propenamide *E*-**6c**, *Z*-**7c**. Total yield 90%. **6c** (pale yellow oil): IR (Nujol) 3100–3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃) 0.80–1.30 (m, 6H, CH₃); 3.00–

3.60 (m, 4H, CH₂); 3.48 (s, 3H, OCH₃); 3.70 (s, 3H, OCH₃); 4.10–4.90 (bs, 1H, NH, exch.); 6.62 (d, AB system, *J* 9 Hz, 2H, aryl-H), 6.94 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.09–7.18 (m, 2H, pyridyl-H); 7.48 (dt, *J*_{ortho} 7.7, *J*_{meta} 1.7 Hz, 1H); 8.63 (dd, *J*_{ortho} 4.4, *J*_{meta} 1.7 Hz, 1H). ¹³C-NMR (CDCl₃): 55.49, 58.67, 114.03, 123.40, 125.50, 126.85, 127.83, 130.78, 136.40, 150.25, 151.02, 153.65, 158.98, 165.22. Calcd. for C₂₀H₂₅N₃O₂ (339.44): C 70.77, H 7.42, N 12.38 found C 70.64, H 7.27, N 12.10. *m/z* 339 (M⁺). HPLC: E isomer *r. t.* 5.65 min. **7c** was obtained only in trace.

3-Cyano-N,N-diethylamino-3-methoxy-2-(4-methoxyphenyl)-3-propenamides E-6d, Z-7d. Total yield 70%. **7d** (white crystals): M.p. 127 °C (Et₂O). IR (Nujol) 3100–3600 cm⁻¹ (NH); 2200 cm⁻¹ (CN). ¹H-NMR (CDCl₃): 0.88–1.20 (m, 6H, CH₃); 3.11–3.46 (m, 4H, CH₂); 3.81 (s, 3H, OCH₃); 3.83 (s, 3H, OCH₃); 5.60–6.20 (bs, 1H, NH, exch.); 6.92 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.54 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 14.00, 42.50, 55.78, 59.51, 114.45, 114.86, 123.89, 125.33, 129.86, 136.94, 161.20, 161.44. Calcd. for C₁₆H₂₁N₃O₂ (287.36): C 66.88, H 7.37, N 14.62 found C 67.04, H 7.17, N 14.20. **6d** (mixture with **7d**): ¹H-NMR (CDCl₃): 0.80–1.20 (m, 6H, CH₃); 3.00–3.50 (m, 4H, CH₂); 3.65 (s, 3H, OCH₃); 3.73 (s, 3H, OCH₃); 6.78 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.29 (d, AB system, *J* 9 Hz, 2H, aryl-H).

Synthesis of Propenamides **9a-c**:

Compound **8** (373 mg, 1 mmol) was dissolved in the appropriate alcohol (15 mL) and sodium alkoxide **2a-c**, (5 mL of a 0.4 molar solution) was added under stirring at room temperature. When the reagent disappeared (T.L.C. cyclohexane/ethyl acetate 1:1; CH₂Cl₂/MeOH 10:1) water (10 mL) was added and stirring continued for 1 h. Alcohol evaporation and extraction of the mixture with CH₂Cl₂ (20 mL) afforded compounds **9a-c** as pale yellow oils.

N,N-Diethylamino-3,3-dimethoxy-2-(4-methoxyphenyl)-propenamidine 9a: Yield: 81%; IR (Nujol) 3200–3400 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 1.05 (t, 6H, CH₃, *J* 7.2 Hz); 3.20–3.50 (m, 4H, CH₂); 3.68 (s, 3H, OCH₃); 3.70 (s, 3H, OCH₃); 3.80 (s, 3H, OCH₃); 6.15–6.40 (bs, 1H, NH, exch.); 6.84 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.36 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 11.70, 13.09, 44.06, 55.60, 56.79, 61.55, 90.13, 114.7, 125.13, 129.98, 158.93, 162.18, 163.36. Calcd. for C₁₆H₂₄N₂O₃ (292.38): C 65.73, H 8.27, N 9.58 found C 65.64, H 8.37, N 9.20. *m/z* 292 (M⁺).

N,N-Diethylamino-3,3-diethoxy-2-(4-methoxyphenyl)-propenamidine 9b: Yield: 76%; IR (Nujol) 3200–3400 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 0.80–1.20 (m, 6H, CH₃); 1.29 (t, 6H, CH₃, *J* 7.2 Hz); 3.00–3.80 (m, 4H, CH₂); 3.79 (s, 3H, OCH₃); 3.93 (q, 4H, OCH₂, *J* 7.2 Hz); 5.50–6.20 (bs, 1H, NH, exch.); 6.84 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.42 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 13.00, 14.98, 43.92, 55.25, 65.44, 68.14, 94.94, 114.06, 125.98, 129.34, 158.14, 158.20, 163.68. Calcd. for C₁₈H₂₈N₂O₃ (320.43): C 67.47, H 8.81, N 8.74 found C 67.67, H 8.58, N 8.95. *m/z* 320 (M⁺).

N,N-Diethylamino-3,3-diisopropoxy-2-(4-methoxyphenyl)-propenamidine 9c: Yield: 70%; IR (Nujol) 3200–3400 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 0.90–1.20 (m, 6H, CH₃); 1.22 (d, CH₃, *J* 6.2 Hz, 6H); 1.25 (d, CH₃, *J* 6.2 Hz, 6H); 3.20–3.60 (m, 4H, CH₂); 3.79 (s, 3H, OCH₃); 4.34 (sept, CH, *J* 6.2 Hz, 2H); 4.40–5.50 (bs, 1H, NH, exch.); 6.81 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.49 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR

(CDCl₃): 22.40, 22.50, 55.53, 72.40, 72.58, 105.60, 113.76, 129.42, 127.97, 151.93, 157.91, 165.19. Calcd. for C₂₀H₃₂N₂O₃ (342.49): C 68.93, H 9.26, N 8.04 found C 68.67, H 8.97, N 8.36. *m/z* 342 (M⁺).

Synthesis of Acrylamidines 11a-c and 12a-f:

Compound **1a** (0.1 g, 0.34 mmol) was dissolved in anhydrous THF (2.0 mL) under stirring at room temperature and Grignard reagents **10a-f** (0.71 mmol) were added dropwise to the solution under nitrogen atmosphere. Stirring was continued until the reagents disappeared (T.L.C. cyclohexane/ethyl acetate 1:1; CH₂Cl₂/MeOH 10:1). Water (5 mL) was added and stirring continued for 1 h. Evaporation of the solvent and extraction of the mixture with CH₂Cl₂ (20 mL) afforded compounds **11** and **12** as yellow oils except in the case of **12b** which crystallized with Et₂O on long standing from the mixture of the two isomers.

N,N-Diethylamino-2-(4-methoxyphenyl)-2-butenamidines **Z-11a**, **E-12a**. Total yield: 73%; IR (Nujol) 3100-3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 0.99 (t, CH₃, *J* 7 Hz, 3H); 1.15-1.40 (m, 6H, CH₃); 1.89 (d, 3-CH₃, *J* 7 Hz, 3H, *Z* form), 1.92 (d, 3-CH₃, *J* 7 Hz, 3H, *E* form), 3.05-3.30 (m, 4H, CH₂); 3.44 (q, CH₂, *J* 7 Hz, 4H), 3.80 (s, 3H, OCH₃); 3.82 (s, 3H, OCH₃); 4.80-5.00 (bs, 1H, NH, exch.); 6.15 (q, 3-H, *J* 7 Hz, 1H, *Z* form); 6.33 (q, 3-H, *J* 7 Hz, 1H, *E* form); 6.86 (d, AB system, *J* 9 Hz, 2H, aryl-H); 6.90 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.21 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.26 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 12.40, 15.30, 42.80, 55.65, 114.42, 122.15, 126.92, 129.99, 130.40, 139.58, 159.54, 165.35, 168.66. Calcd. for C₁₅H₂₂N₂O (246.35): C 73.13, H 9.00, N 11.37 found C 73.00, H 8.77, N 11.05. *m/z* 246 (M⁺). HPLC: *E* isomer r. t. 4.65 min., *Z* isomer r. t. 5.49 min.

N,N-Diethylamino-2-(4-methoxyphenyl)-2-pentenamidines **Z-11b**, **E-12b**. Total yield: 98%; **11b** (pale yellow crystals on long standing): M.p. 194-196 °C dec. (Et₂O). IR (Nujol) 3100-3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 0.98 (t, CH₃, *J* 7 Hz, 3H); 1.15 (t, CH₃, *J* 7 Hz, 3H); 1.43 (t, CH₃, *J* 7 Hz, 3H); 2.17-2.35, 2.38-2.54 (2m, 2H, CH₂); 3.25-3.30, (m, 2H, CH₂); 3.80 (s, 3H, OCH₃); 3.81-4.01 (m, 2H, CH₂); 5.00-5.50 (bs, 1H, NH, exch.); 6.21 (t, 3-H, *J* 7.7 Hz, 1H); 6.87 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.30 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 12.78, 14.30, 23.94, 44.15, 55.74, 114.60, 127.34, 129.36, 130.37, 136.29, 138.00, 159.96, 164.78. Calcd. for C₁₆H₂₄N₂O (260.38): C 73.81, H 9.29, N 10.76 found C 74.00, H 9.37, N 10.95. *m/z* 260 (M⁺). **12b** ¹H-NMR (CDCl₃): 0.80-1.40 (m, 9H, CH₃); 2.10-2.40 (m, 2H, CH₂); 2.95-3.65 (m, 4H, CH₂); 3.80 (s, 3H, OCH₃); 5.97 (t, 3-H, *J* 7.7 Hz, 1H); 6.90 (d, AB system, *J* 9 Hz, 2H, aryl-H); 7.20 (d, AB system, *J* 9 Hz, 2H, aryl-H). ¹³C-NMR (CDCl₃): 12.00-14.00, 22.71, 44.15, 55.27, 114.60, 130.30, 131.20, 134.82, 136.00, 167.08. *m/z* 260 (M⁺).

N,N-Diethylamino-2-(4-methoxyphenyl)-3-phenyl-propenamidines **Z-11c**, **E-12c**. Total yield: 93%; IR (Nujol) 3100-3600 cm⁻¹ (NH). ¹H-NMR (CDCl₃): 0.80 (t, CH₃, *J* 7 Hz, 3H, *Z* form); 1.04 (t, CH₃, *J* 7 Hz, 6H, *E* form); 1.34 (t, CH₃, *J* 7 Hz, 3H, *Z* form); 3.18-3.24 (m, 1H, CH₂ *Z* form); 3.39 (q, CH₂, *J* 7 Hz, 4H, *E* form); 3.65-3.83 (m, 3H, CH₂ *Z* form); 3.86 (s, 3H, OCH₃ *E* form); 3.89 (s, 3H, OCH₃ *Z* form); 5.40-5.80 (bs, 1H, NH, exch.); 6.70 (s, 1H, H-3, *E* form); 6.80 (d, AB system, *J* 9 Hz, 2H, aryl-H, *E* form); 6.93 (AB system, *J* 9 Hz, 2H, aryl-H, *Z* form); 7.50 (AB system, *J* 9 Hz, 2H, aryl-H, *Z* form); 7.11-7.73 (2m, 13H, aryl-H). ¹³C-NMR (CDCl₃): 12.96, 42.14, 55.25, 114.06, 126.60, 127.80, 128.20, 128.26, 128.49, 129.32,

136.29, 140.47, 159.47, 168.90. Calcd. for $C_{20}H_{24}N_2O$ (308.42): C 77.89, H 7.84, N 9.08 found C 77.67, H 7.97, N 9.45. m/z 308 (M^+).

N,N-Diethylamino-2-(4-methoxyphenyl)-pent-2-en-4-ynamidine **E-12d**. Yield: 80%; IR (Nujol) 3100–3600 cm^{-1} (NH). 1H -NMR ($CDCl_3$): 0.90 (t, CH_3 , J 7 Hz, 6H); 3.18 (q, CH_2 , J 7 Hz, 4H); 3.22 (d, J 2.65 Hz, 1H); 3.78 (s, 3H, OCH_3); 5.67 (d, J 2.65 Hz, 1H); 5.75 (bs, 1H, NH, exch.); 6.87 (d, AB system, J 9 Hz, 2H, aryl-H); 7.69 (d, AB system, J 9 Hz, 2H, aryl-H). ^{13}C -NMR ($CDCl_3$): 13.00, 41.90, 42.04, 55.27, 84.26, 94.25, 105.00, 113.80, 127.13, 129.80, 148.38, 160.25, 166.90. Calcd. for $C_{16}H_{20}N_2O$ (256.35): C 74.97, H 7.26, N 10.93 found C 74.67, H 7.17, N 10.85. m/z 256 (M^+).

N,N-Diethylamino-2-(4-methoxyphenyl)-hex-2-en-4-ynamidine **E-12e**. Total yield: 90%; IR (Nujol) 3100–3600 cm^{-1} (NH). 1H -NMR ($CDCl_3$): 0.95 (t, CH_3 , J 7 Hz, 6H); 1.96 (d, CH_3 , J 2.65 Hz, 3H); 3.27 (q, CH_2 , J 7 Hz, 4H); 3.76 (s, 3H, OCH_3); 5.71 (q, J 2.65 Hz, 1H); 6.40 (bs, 1H, NH, exch.); 6.85 (d, AB system, J 9 Hz, 2H, aryl-H); 7.65 (d, AB system, J 9 Hz, 2H, aryl-H). ^{13}C -NMR ($CDCl_3$): 4.73, 12.63, 42.89, 55.25, 70.00, 94.91, 108.83, 114.00, 126.84, 129.07, 143.89, 160.01, 162.24. Calcd. for $C_{17}H_{22}N_2O$ (270.37): C 75.52, H 8.20, N 10.36 found C 75.67, H 8.10, N 10.57. m/z 270 (M^+).

N,N-Diethylamino-2-(4-methoxyphenyl)-5-phenyl-pent-2-en-4-ynamidine **E-12f**. Total yield: 94%; IR (Nujol) 3100–3600 cm^{-1} (NH). 1H -NMR ($CDCl_3$): 1.05 (t, CH_3 , J 7 Hz, 6H); 3.29 (q, CH_2 , J 7 Hz, 4H); 3.85 (s, 3H, OCH_3); 5.96 (s, 1H, H-3); 6.93 (d, AB system, J 9 Hz, 2H, aryl-H); 7.30–7.55 (m, 5H, aryl-H); 7.82 (d, AB system, J 9 Hz, 2H, aryl-H). ^{13}C -NMR ($CDCl_3$): 13.46, 42.36, 55.78, 88.11, 96.51, 106.37, 114.18, 128.08–132.96 (CH arom.), 122.33, 123.73, 148.75, 160.62, 167.66. Calcd. for $C_{22}H_{24}N_2O$ (332.45): C 79.48, H 7.28, N 8.43 found C 79.77, H 7.18, N 8.59. m/z 332 (M^+).

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